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11—1 Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is $Mg + 2HCl \rightarrow MgCl_2 + H_2$ 14. Aluminum reacts with HCl to produce aluminum chloride ($AlCl_3$) and hydrogen ...

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10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compounds. Show all your work. 11. $CaCO_3$ 12. $MgSO_4$ 13. $NaOH$ 14. KCl 15. BaF_2 16. Na_2CO_3 17. Na_2SO_4 18. $CaSO_4$ 19. K_2SO_4 20. Na_2O 21. Na_2S 22. Na_2O_2 23. Na_2O_3 24. Na_2O_4 25. Na_2O_5 26. Na_2O_6 27. Na_2O_7 28. Na_2O_8 29. Na_2O_9 30. Na_2O_{10} 31. Na_2O_{11} 32. Na_2O_{12} 33. Na_2O_{13} 34. Na_2O_{14} 35. Na_2O_{15} 36. Na_2O_{16} 37. Na_2O_{17} 38. Na_2O_{18} 39. Na_2O_{19} 40. Na_2O_{20} 41. Na_2O_{21} 42. Na_2O_{22} 43. Na_2O_{23} 44. Na_2O_{24} 45. Na_2O_{25} 46. Na_2O_{26} 47. Na_2O_{27} 48. Na_2O_{28} 49. Na_2O_{29} 50. Na_2O_{30} 51. Na_2O_{31} 52. Na_2O_{32} 53. Na_2O_{33} 54. Na_2O_{34} 55. Na_2O_{35} 56. Na_2O_{36} 57. Na_2O_{37} 58. Na_2O_{38} 59. Na_2O_{39} 60. Na_2O_{40} 61. Na_2O_{41} 62. Na_2O_{42} 63. Na_2O_{43} 64. Na_2O_{44} 65. Na_2O_{45} 66. Na_2O_{46} 67. Na_2O_{47} 68. Na_2O_{48} 69. Na_2O_{49} 70. Na_2O_{50} 71. Na_2O_{51} 72. Na_2O_{52} 73. 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