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Write a balanced equation describing each of the following chemical reactions. (a) Solid potassium chlorate, KClO<sub>3</sub>, decomposes to form solid potassium chloride and diatomic oxygen gas. (b) Solid aluminum metal reacts with solid diatomic iodine to form solid Al<sub>2</sub>I<sub>6</sub>. (c) When solid sodium chloride is added to aqueous sulfuric acid, hydrogen chloride gas and aqueous sodium sulfate are produced.

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**Chemistry Form 4 Chapter 3 Exercise And Answer**  
Topical test chapter 3 Chemistry Form 4 1. Name : \_\_\_\_ Class : \_\_\_\_ TOPICAL TEST CHAPTER 3 1. Write down the chemical equation of the following reactions a. Sodium reacts with oxygen to produce a white metallic oxide.

**Topical test chapter 3 Chemistry Form 4 - SlideShare**  
Element Relative mass Mass of 1 mol Molar mass Helium 4 4 4 g mol<sup>-1</sup> Sodium 23 23 23 g mol<sup>-1</sup> Water, H<sub>2</sub>O 2(1) + 16 = 18 18 18 g mol<sup>-1</sup> Ammonia, NH<sub>3</sub> 14 + 3(1) = 17 17 17 g mol<sup>-1</sup> 17. Relationship between the number of moles and the mass of a substance Number of moles, (mol) Mass (g) × molar mass + molar mass 18.

**Chapter 3 Chemical Formulae and Equations**  
3PO<sub>4</sub> Exercise 5.13 - Making solution from Solid: An experiment calls for a total of 1.50 L of 0.200 M KMnO<sub>4</sub> for a class of chemistry students. How would this solution be made from pure, solid potassium permanganate and water? 44 4 4 0.200 mol KMnO 158.0339 g KMnO<sub>7</sub> g KMnO = 1.50 L soln 1 L soln 1 mol KMnO ...

**Chapter 2 Exercise Key - Home - Faculty**  
Exercise and physical activity are great ways to feel better, boost your health and have fun. For most healthy adults, the Department of Health and Human Services recommends: At least 150 minutes a week of moderate aerobic activity or 75 minutes a week of vigorous aerobic activity, or a combination of moderate and vigorous activity.