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In chemistry, the standard state of a material (pure substance, mixture or solution) is a reference point used to calculate its properties under different conditions. A superscript circle is used to designate a thermodynamic quantity in the standard state, such as change in enthalpy ( $\Delta H^\circ$ ), change in entropy ( $\Delta S^\circ$ ),

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or change in Gibbs free energy ( $\Delta G^\circ$ ).

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