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and Kriz to help with revisions to the text. Professor Vyvyan's areas of interests include the total synthesis of natural products, development of synthetic methods, and structure determination using NMR.

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Pavia, Gary M.
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the dispersion of light
into component colors.
In simple words, it is a
method to measure
how much light is
absorbed by a
chemical substance
and at what intensity
of light passes through
it.

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changes. The discussion of coupling constant analysis in Chapter 5 has been significantly expanded. Long-range couplings are covered in more detail, and multiple strategies for measuring coupling constants are presented.

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of INTRODUCTION TO SPECTROSCOPY, he joined the author team with Pavia, Lampman, and Kriz to help with revisions to the text. Professor Vyvyan's areas of interests include the total synthesis of natural products, development of synthetic methods, and structure determination using NMR.

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90.50% carbon; 9.50%
hydrogen (b) C₄H₅ 2.
32.0% carbon; 5.4%
hydrogen; 62.8%
chlorine; C₃H₆Cl₂ 3.
C₂H₅NO₂ 4. 180.2 =
molecular mass.
Molecular formula is C
9H₈O₄ 5. Equivalent

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Donald L. Pavia Gary
M. Lampman George S.
Kriz James R. Vyvyan.
Department of
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