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Student Exploration Half Life Gizmo

You can use the Half-life Gizmo to model the decay of Carbon-14, which has a half-life of approximately 6,000 years (actual value is 5,730 years). In the Gizmo, select User chooses half-life and Theoretical decay. Set the Half-life to 6

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seconds (to represent 6,000 years) and the Number of atoms to 100.

Student Exploration: Half-life (ANSWER KEY)

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Theoretical decay. Set the Half-life to 6 seconds (to represent 6,000 years) and the Number of atoms to 100.

Student Exploration: Half-life - Kent City School District

Investigate the decay of a radioactive substance. The half-life and the number of radioactive atoms can be adjusted, and theoretical or random

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decay can be observed. Data can be interpreted visually using a dynamic graph, a bar chart, and a table. Determine the half-lives of two sample isotopes as well as samples with randomly generated half-lives.

Half-life Gizmo : ExploreLearning

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Investigate the decay of a radioactive substance. The half-life

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Radioactive atoms change by emitting radiation in the form of tiny particles and/or energy. This process, called decay, causes the radioactive atom to change into a stable daughter atom. The Half-life Gizmo™ allows you to observe and measure the decay of a radioactive substance. Be sure the sound is turned on and click Play ().

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6,000 years) and the
Number of atoms to
100.

Student Exploration Half Life Gizmo Answer Key

Get the Gizmo ready:
Click Reset, and select
the Micro view. Set Ice
volume to 0 cc. Set
Add/remove heat
energy to 0 J/s.

Question: Why do
phase changes occur?

1. Compare: Set the
Water temperature to

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0 °C and click Play.

Observe the water
molecules. Click Reset,
set the Water
temperature to 100 °C,
and click Play again.

**Get the Gizmo
ready: Activity B:
Reset Micro view ...**

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Student Exploration Sheet: Growing Plants

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Elements like C14 and
U238 are used for
absolute dating of
fossils and rocks.
These radioactive
elements break down
(decay) at a steady
rate into “daughter
atoms.” We can
compare how many of
the

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